Metabolism of the crucifer phytoalexins wasalexin A and B in the plant pathogenic fungus *Leptosphaeria maculans*[†]

M. Soledade C. Pedras* and Mojmír Suchý

Received 3rd July 2006, Accepted 19th July 2006 First published as an Advance Article on the web 14th August 2006 DOI: 10.1039/b609367a

Wasalexins A and B are crucifer phytoalexins produced by two substantially different plant species, a wild species abundant in the Canadian prairies and a condiment plant widely cultivated in Japan. Interestingly, both plant species are resistant to an economically important fungal plant pathogen, the blackleg fungus [*Leptosphaeria maculans* (Desm.) Ces. et de Not., asexual stage *Phoma lingam* (Tode ex Fr.) Desm.]. The transformation of wasalexins A and B in cultures of isolates of *L. maculans*, an isolate highly virulent towards canola (BJ 125) and a less common isolate which is virulent towards wasabi (Laird 2/Mayfair 2) was investigated. It was established that both fungal isolates are able to efficiently metabolize and detoxify wasalexins A and B through reduction in the case of wasalexin A or through hydrolysis followed by reduction in the case of wasalexin B. Moreover, a close structural analogue of wasalexins, which does not occur naturally, was also found to be reduced in cultures of *L. maculans*. The structures of the new metabolic products were elucidated using spectroscopic methods and were confirmed by synthesis. Bioassays indicated that the biotransformation of wasalexins is a detoxification process that may contribute to the aggressive nature of these fungal isolates towards plants that produce wasalexins.

Introduction

The phytoalexins wasalexins A (1a) and B (1b) are produced in response to biotic and abiotic stress in wasabi (Wasabia japonica syn. Eutrema wasabi), a condiment plant widely cultivated in Japan,¹ and in pennycress or stinkweed (Thlaspi arvense), a weed common in the Canadian prairies.² Phytoalexins are chemical defenses biosynthesized de novo by plants in response to pathogen attack and other forms of stress. Because wasalexins A (1a) and B (1b) were found to be produced in two rather different plant species that are known to be resistant to the fungus Leptosphaeria maculans (Desm.) Ces. et de Not. [asexual stage Phoma lingam (Tode ex Fr.) Desm.], it was of great interest to investigate their effect on this pathogen. L. maculans is the causative agent of "blackleg", an economically important disease of crucifer crops worldwide. This plant pathogen is able to transform enzymatically plant defense metabolites, phytoalexins and related metabolites, to products with significantly lower antifungal activity.³ For example, L. maculans detoxified the phytoalexins brassinin (2),³ cyclobrassinin (4)³ and brassilexin $(5)^4$ to products 3 and 6, respectively (Scheme 1). Detoxification reactions of phytoalexins in plant pathogenic fungi are detrimental to the plant and favourable to the pathogen. Consequently, inhibition of these enzymatic processes could be used to prevent fungal colonization of plant tissues. Toward this goal, we have been evaluating detoxification pathways of crucifer phytoalexins and designing phytoalexin detoxification inhibitors, *i.e.* paldoxins, to protect the plant against fungal colonization.³



Scheme 1 Metabolism of brassinin (2),³ cyclobrassinin $(4)^3$ and brassilexin $(5)^4$ in *Leptosphaeria maculans*.

In a continuation of those studies, we investigated the transformation of wasalexins A (1a) and B (1b) in two different isolates of *L. maculans*, an isolate that is highly virulent towards canola

Department of Chemistry, University of Saskatchewan, 110 Science Place, Saskatoon, SK, Canada S7N 5C9. E-mail: s.pedras@usask.ca Fax: +1 (1)306 9664730; Tel: +1 (1)306 9664772

[†] Electronic supplementary information (ESI) available: ¹H NMR of new compounds for proof of purity. See DOI: 10.1039/b609367a

Table 1 Metabolism of wasalexin A (1a) and its metabolites 7 and 8 in cultures of Leptosphaeria maculans isolates BJ 125 and Laird 2/Mayfair 2

Compound added to fungal cultures	Products of metabolism (%) ^a after incubation for 6–96 h	
	Isolate BJ 125	Isolate Mayfair 2
Wasalexin A (1a) ^b	6 h: 7 (47 ± 3); 8 (46 ± 1)	6 h: 7 (28 \pm 8); 8 (18 \pm 2)
Dihydrowasalexin $(7)^b$	6 h: 8 (49 \pm 3)	6 h: 7 (18 \pm 2); 8 (24 \pm 4);
•	12 h: 8 (66 \pm 1)	12 h: $7 (<5)$; $8 (21 \pm 7)$
	24 h: 8 (78 \pm 3)	24 h: 8 (31 \pm 6)
1-Methoxy-3-methylindol-2-one (8) ^c	Slow transformation to undetermined products (79 \pm 6 after 96 h)	Slow transformation to undetermined products (38 \pm 5 after 96 h)

^{*a*} Average percentage yield relative to total amount of compound added to culture (% \pm standard deviations; triplicate samples) calculated by HPLC analysis using calibration curves constructed for each compound. ^{*b*} Compound (5 × 10⁻⁵ M concentration in the culture) dissolved in DMSO was added to 7-d-old mycelia in water and was incubated at 24 \pm 2 °C. ^{*c*} Compound (5 × 10⁻⁵ M concentration in the culture) dissolved in DMSO was added to 48-h-old mycelia in minimal media and was incubated at 24 \pm 2 °C.

(BJ 125) and a less common isolate which is virulent towards wasabi (Laird 2/Mayfair 2).⁵ It was established that both isolates are able to efficiently metabolize and detoxify wasalexins using either two sequential reductive steps in the case of wasalexin A (1a) or hydrolysis followed by reduction in the case of wasalexin B (1b). Moreover, a close structural analogue of wasalexins, which does not occur naturally, was also found to be reduced in both isolates of *L. maculans.* The structures of the new products were determined by spectroscopic methods and confirmed by total synthesis. The results of these studies are described here for the first time.

Results and discussion

Fungal metabolism

Wasalexins A (1a) and B (1b) were synthesized and obtained as a mixture of E (wasalexin A) and Z (wasalexin B) stereoisomers in a 2 : 1 ratio, as described previously.⁶ The more stable E isomer (1a) could be obtained by crystallization of the mixture using a dichloromethane-hexane solution.² The less stable Z isomer (1b) could not be obtained as a single compound.1 The metabolism of wasalexins A and B (5 \times 10⁻⁵ M) in cultures of L. maculans was analyzed over a period of several days by HPLC. The HPLC chromatograms of EtOAc extracts of cultures of L. maculans, isolates BJ 125 and Mayfair 2 (mycelia in water) incubated with wasalexin A (1a) (t_R 25.2 min) showed two new peaks, at t_R 10.3 min and $t_{\rm R}$ 23.7 min, not present in control cultures (no wasalexins). Furthermore, wasalexin A (1a) was not detected in extracts of cultures of isolates BJ 125 or Mayfair 2 after incubation for six hours; that is **1a** appeared to be completely metabolized. Subsequently, from larger scale cultures, the metabolites corresponding to the new peaks at $t_{\rm R}$ 10.3 min and $t_{\rm R}$ 23.7 min were isolated and purified (Table 1). The metabolite with $t_{\rm R}$ 10.3 min (8) was found to have the molecular formula $C_{10}H_{11}NO_2$ (HRMS-EI). The ¹H NMR spectrum (see the ESI[†]) displayed signals due to four aromatic protons and signals at δ 3.45 (J = 8 Hz, q, 2H) and δ 1.52 (J = 8 Hz, d, 3H) indicating the presence of a methine-methyl spin system. This metabolite was proposed to be 1-methoxy-3-methylindol-2-one (8), the structure of which was confirmed by synthesis as described in the Experimental section. Similarly, analysis of the spectroscopic data obtained for the metabolite with $t_{\rm R}$ 23.7 min and comparison with those of wasalexin A (1a) suggested it to be dihydrowasalexin (7). Synthesis of this metabolite confirmed its structure unambiguously. Subsequent incubation of dihydrowasalexin (7) in cultures of *L. maculans* isolates BJ 125 and Mayfair 2 indicated it to be the precursor of oxindole **8** (Scheme 2, Table 1). Oxindole **8** was slowly biotransformed (in more than 96 h) to undetermined products. As shown in Scheme 2, wasalexin A was first reduced at the exocyclic double bond to yield dihydrowasalexin (7), followed by further reduction of the methylene to a methyl to yield **8**. These metabolism experiments (Scheme 2, Table 1) were carried out with mycelia incubated in water because one of the metabolites of wasalexin A, dihydrowasalexin (7), was found to decompose non-enzymatically in minimal media (<5% of 7 remaining in solution after 12 h) with the formation of 1-methoxy-3-methylsulfanylmethylindol-2-one (**9**). Further studies to understand the formation of **9** were carried out and are described below.



Scheme 2 Metabolism of wasalexin A (1a) in *Leptosphaeria maculans* isolates BJ 125 and Laird 2/Mayfair 2.

Next, because wasalexin B (1b) could not be obtained as a single compound (on standing in solution it is converted to wasalexin A (1a)), a mixture of wasalexins A and B (1a + 1b, 2 : 1) was incubated with cultures of *L. maculans* and their metabolism analyzed over a period of several days. HPLC analyses of the EtOAc extracts of cultures of *L. maculans* isolate BJ 125 (48-h-old mycelia in minimal media) incubated with wasalexins A and B (1a + 1b, 2 : 1) showed the disappearance of both wasalexins peaks (t_R 22.6 and 25.2 min) within 12 h. Dihydrowasalexin (7, t_R 23.7 min) and

 Table 2
 Metabolism of wasalexins A + B (1a + 1b, 2: 1) and metabolites 10–12 (metabolism of compounds 7 and 8 shown in Table 1) in cultures of Leptosphaeria maculans isolates BJ 125 and Laird 2/Mayfair 2

Products of metabolism (%) ^a after incubation for 6–96 h		
Isolate BJ 125	Isolate Mayfair 2	
6 h: 7 (15 ± 1); 8 (28 ± 4)	6 h: 7 (28 ± 8); 8 (18 ± 2)	
10 (9 ± 1) ; 11 (< 5)	12 h: 7 (15 \pm 1); 8 (31 \pm 1); 11 (8 \pm 4)	
12 h: 11 (16 \pm 3)	12 h: 11 (<5); 12 (<5)	
24 h: 11 (62 \pm 7)	24 h: 11 ($<$ 5); 12 (5 \pm 2)	
48 h: 11 (64 \pm 5)	48 h: 11 (7 \pm 1); 12 (18 \pm 4)	
	96 h: 11 (9 \pm 1): 12 (42 \pm 7)	
Slow transformation to undetermined	Slow transformation to undetermined	
products $(31 \pm 3 \text{ recovered after } 96 \text{ h})$	products $(12 \pm 2 \text{ recovered after } 96 \text{ h})$	
Not formed	Slow transformation to undetermined products $(57 \pm 5 \text{ after } 96 \text{ h})$	
	Products of metabolism (%) ^a after incubation for the second s	

^{*a*} Average percentage yield ($\% \pm$ standard deviations; triplicate samples) was calculated by HPLC analysis using calibration curves constructed for each compound. ^{*b*} Compounds (5 × 10⁻⁵ M concentration in the culture) dissolved in DMSO were added to 7-d-old cultures in water and were incubated at 24 ± 2 °C. ^{*c*} Compounds (5 × 10⁻⁵ M concentration in the culture) dissolved in DMSO were added to 48-h-old cultures in minimal media and were incubated at 24 ± 2 °C.

1-methoxyindol-2-one (8, t_R 10.3 min) together with two additional metabolites with $t_{\rm R}$ 9.0 and 20.6 min were detected in the HPLC chromatograms of extracts of cultures. The presence of additional metabolites in cultures of isolate BJ 125 incubated with the mixture of wasalexins A and B suggested that the metabolism of wasalexin B (1b) was different from the metabolism of wasalexin A (1a). Subsequently, larger scale experiments allowed the isolation of the metabolites responsible for peaks with $t_{\rm R}$ 9.0 and 20.6 min in the HPLC chromatograms. The HRMS-EI data of the metabolite with t_R 20.6 min (10) suggested the molecular formula C₁₂H₁₂N₂O₃S, *i.e.* one less SCH₃ and an additional oxygen relative to wasalexin. Comparison of the ¹H and ¹³C NMR spectral data of this metabolite with those of wasalexins indicated this metabolite to be thiocarbamate 10 (Scheme 3), the structure of which was confirmed by synthesis. Thiocarbamate 10 was found to be biotransformed efficiently to 11 in cultures of BJ 125 (Table 2, Scheme 3). The HRMS-EI spectral data of the metabolite with $t_{\rm R}$ 9.0 min (11) suggested the molecular formula $C_{12}H_{14}N_2O_3S$. The ¹H NMR spectrum displayed signals due to four aromatic protons, singlets at δ 4.05 ppm (3H) and δ 2.37 (3H) indicating the presence of OCH₃ and SCH₃ groups, respectively, and multiplets at δ 4.19 (1H), 3.62 (1H) and 3.46 (1H) ppm. Furthermore, the two lowest



Scheme 3 Metabolism of wasalexin B (1b) in *Leptosphaeria maculans* isolates BJ 125 and Laird 2/Mayfair 2.

field signals in the ¹³C NMR spectrum suggested the presence of two carbonyls likely due to amide and (thio)carbamate groups. Thus, the NMR data of the metabolite with t_R 9.0 min (11) indicated that the double bond present in the side chain of both wasalexin B and metabolite 10 had been reduced whereas the thiocarbamate remained intact. The compound was proposed to be *S*-methyl 1-methoxy-3-aminomethyl-2-oxindole thiocarbamate (11). This structure was confirmed by synthesis as described in the Experimental section.

HPLC analysis of the EtOAc extracts obtained from cultures of L. maculans isolate Mayfair 2 (48-h-old mycelia in minimal media) incubated with wasalexins A and B (1a + 1b, 2 : 1) (Table 2) indicated the presence of dihydrowasalexin (7), oxindole 8 and thiocarbamate 11 but metabolite 10 was not detected. The absence of 10 in these chromatograms suggested that either wasalexin B (1b) was transformed directly to thiocarbamate 11 or that 10 did not accumulate because it was immediately transformed to 11, *i.e.* the rate of transformation of 10 is similar to its rate of formation (Scheme 3). Furthermore, dihydrowasalexin (7) was not a likely intermediate in the formation of 11 because this metabolite was not detected when 7 was incubated in cultures of isolate Mayfair 2. Considering that in cultures of isolate Mayfair 2 metabolite 10 was found to be metabolized to thiocarbamate 11 and to 1-methoxy-3aminomethyleneindol-2-one (12)² (Table 2), 10 is a likely precursor of 11.

As summarized in Scheme 3, the metabolism of wasalexin B in isolates BJ 125 and Laird 2/Mayfair 2 yields 11 likely *via* intermediate 10. However, in cultures of isolate Laird 2/Mayfair 2 metabolite 10 appears to be transformed mainly to enamine 12, whereas in BJ 125 10 yields mainly metabolite 11. Such differences are not unusual considering that these two isolates are part of different pathogenicity groups (currently classified as belonging to the same species).⁵

Next, to further understand the enzymatic reactions leading to the transformation of wasalexins A (1a) and B (1b), 1methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13) was synthesized² and its transformation in cultures of *L. maculans* was investigated. When methyl 1-methoxy-3-aminomethylene-2oxindole dithiocarbamate (13) was incubated with isolate BJ 125

Table 3Metabolism of methyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13) and its metabolites 14 and 15 (metabolism of 12 in
Table 2) in cultures of Leptosphaeria maculans isolates BJ 125 and Mayfair 2

	Products of metabolism (%) ^a after incubation for 6–96 h		
Compound added to fungal cultures	Isolate BJ 125	Isolate Mayfair 2	
Methyl 1-methoxy-3-	12 h: 14 (12 ± 1); 15 (<5)	12 h: 15 (<5)	
aminomethylene-2-oxindole	24 h: 14 (29 \pm 2); 15 (<5)	24 h: 12 (<5); 15 (<5)	
dithiocarbamate $(13)^b$	48 h: 12 (<5); 14 (29 \pm 1); 15 (<5)	48 h: 8 (<5); 12 (9 \pm 1); 14 (5 \pm 1); 15 (<5)	
		72 h: 8 (<5); 12 (14 \pm 1); 14 (<5); 15 (<5)	
Methyl	12 h: 15 (<5)	12 h: 15 (<5)	
1-methoxy-3-aminomethyl-2-oxindole	24 h: 8 (<5); 15 (<5)	24 h: 15 (<5)	
dithiocarbamate $(14)^{b}$	48 h: 8 (7 \pm 2); 15 (6 \pm 1)	48 h: 8 (11 \pm 1); 15 (<5)	
	96 h: 8 (29 \pm 9); 15 (6 \pm 1)	96 h: 8 (13 \pm 1); 15 (<5)	
1-Methoxyspirobrassinin (15) ^b	Slow transformation to undetermined	Slow transformation to undetermined products	
	products (96 h, 49 \pm 8)	$(96 \text{ h}, 34 \pm 5)$	

^{*a*} Average percentage yield (% \pm standard deviations; triplicate samples) was calculated by HPLC analysis using calibration curves constructed for each compound. ^{*b*} Compounds (5 × 10⁻⁵ M in the culture) dissolved in DMSO were added to 48-h-old cultures in minimal media and were incubated at 24 \pm 2 °C.

a peak corresponding to a likely new metabolite with $t_{\rm R}$ 18.5 min (14) was observed in the HPLC chromatograms of EtOAc extracts (12-48 h incubations), in addition to metabolite 8 (major, Table 3) and 15 (by comparison with our HPLC-UV library database). Subsequently, a larger scale biotransformation experiment was carried out to isolate the metabolites. HRMS-EI spectral data indicated that the new metabolite with $t_{\rm R}$ 18.5 min (14) had the molecular formula C₁₂H₁₄N₂O₂S₂. The ¹H NMR spectrum of this new metabolite was similar to that of thiocarbamate 11 but a signal at δ 200.2 ppm in the ¹³C NMR spectra (see the ESI[†]) suggested the presence of a thiocarbonyl carbon. Hence the structure of this compound was proposed to be methyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (14) (Scheme 4). This structure was confirmed by synthesis as described in the Experimental section. In addition, the structure of metabolite 15 was confirmed to be 1methoxyspirobrassinin. These results showed that the metabolism of dithiocarbamate 13 in cultures of L. maculans isolate BJ 125 was much slower than that of wasalexin A (1a) (ca. 20% of 13 remaining after 96 h vs complete metabolism of 1a in 6 h) yielding two products, the major product appeared to be 1methoxy-3-methylindol-2-one (8) together with small amount of



Scheme 4 Metabolism of methyl 1-methoxy-3-aminomethylene-2oxindole dithiocarbamate (13) in *Leptosphaeria maculans* isolates BJ 125 and Laird 2/Mayfair 2.

1-methoxyspirobrassinin (15) (Table 3, Scheme 4). Slow oxidation of dithiocarbamate 14 (*ca.* 3% in 96 h) to 1-methoxyspirobrassinin (15) was also observed in control flasks containing dithiocarbamate 14 in minimal medium. 1-Methoxyspirobrassinin (15)⁸ was found to be slowly metabolized in cultures of *L. maculans* isolate BJ 125 to undetermined product(s) (Table 3, *ca.* 49% of 15 remaining after 96 h).

Similarly, the metabolism of dithiocarbamate 13 in cultures of *L. maculans* isolate Mayfair 2 was slower than that of wasalexin A (1a) (*ca.* 72 h *vs.* complete metabolism of 1a in 6 h). Enamine 12 was found to be the major product while dithiocarbamate 14 was a minor metabolite (Table 3, Scheme 4). As in the case of isolate BJ 125, dithiocarbamate 14 was found to be slowly transformed to 1-methoxy-3-methylindol-2-one (8) and 1-methoxyspirobrassinin (15) (Table 3, Scheme 4). 1-Methoxyspirobrassinin (15) was found to be slowly metabolized to undetermined product(s) (*ca.* 34% of 15 remaining after 96 h). Once again, it is worthy to note that in cultures of isolate Mayfair 2 compound 13 appears to be transformed mainly to enamine 12, whereas in cultures of BJ 125 13 yields mainly metabolite 8 (Scheme 4, Table 3). As mentioned in the case of metabolite 10, these differences are not unusual and are perhaps reflecting the substantial differences between these isolates.⁵

Metabolites 7, 8, 11 and 14 containing a stereogenic center were found to be readily epimerizable in an acidic solution of CD₃CN– D₂O (1 : 2, 1 mL containing 10 μ l of 0.1 M DCl). Hence, the low values of the optical rotations of these metabolites are likely due to their racemization in liquid culture (Experimental section, $[a]_D$ ca. -5-(+6)). However, to the best of our knowledge, compounds 7, 8, 11 and 14 are not known, thus the possibility that enantiomerically pure metabolites with low values of $[a]_D$ are products of the fungal metabolism cannot be excluded. Conversely, it cannot be ruled out that the enzyme(s) involved in these conversions are not stereoselective.

Decomposition of dihydrowasalexin (7)

As stated above, dihydrowasalexin (7) was found to decompose in minimal media (<5% of 7 remaining after standing in solution for 12 h) with the formation of 1-methoxy-3-methylsulfanyl-methylindol-2-one (9, *ca.* 30% yield) and additional undetermined

products. The decomposition of dihydrowasalexin (7) in water yielded the same product **9** but was found to be significantly slower (*ca.* 80% of **7** remaining after standing in solution for 24 h). For this reason, additional studies were carried out to understand this nonenzymatic reaction. It was surmised that the SCH₃ group of **9** was likely derived from one of the SCH₃ groups of **7**, thus analogues **16** and **17** were thought to confirm this hypothesis. Compound **16** was prepared, added to minimal medium (5×10^{-5} M) and incubated under similar conditions used for **7** (room temperature on a shaker). HPLC analysis of extracts of the solution containing **16** (<10% of **16** after 12 h) indicated the presence of oxindoles **9** and **18** in an 1 : 1 ratio (*ca.* 30% overall yield, Scheme 5). In addition, compound **17** was prepared and treated similarly to yield **18** and undetermined products (*ca.* 30% overall yield, Scheme 5).



Scheme 5 Decomposition of dihydrowasalexin (7), *S*-ethyl,*S*-methyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (16) and *S*,*S*-diethyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (17) in minimal media.

A rationale for the formation of both compounds **9** and **18** from **16** is proposed in Scheme 6. Elimination of volatiles R¹SCN and R²SCN from **16** could lead to unstable 1-methoxy-3-methyleneindol-2-one (**19**) plus KSR² and KSR¹. Next, since **19** is an α , β -unsaturated amide, Michael addition of sulfides KSR¹ or KSR² (K⁺ from KNO₃ and K₂HPO₄ present in minimal media) could yield stable thioethers **9** and **18**. In fact, the reaction of 3-methyleneindol-2-one,⁸ presumably less reactive than the 1-methoxy analogue **19**, with various thiols was reported.⁹ Nonetheless, the amount of sulfides **9** and **18** formed in these



Scheme 6 Rationale for formation of compounds 9 and 18 from *S*-ethyl,*S*-methyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (16) in minimal media.

reactions is low (*ca.* 30%, Fig. 1) which suggests that intermediate alkene **19** undergoes additional transformation to polar products insoluble in EtOAc and thus not detected or isolated.



Fig. 1 Formation of 1-methoxy-3-methylsulfanylmethylindol-2-one (9) from dihydrowasalexin (7) in minimal media. Concentrations were determined using calibration curves; each point is an average of triplicate samples \pm standard deviation.

Synthesis of new metabolites

The chemical synthesis of the metabolites involved in the biotransformation pathways of wasalexins A (1a) and B (1b) and dithiocarbamate 13 was carried out in order to confirm their structures and to obtain sufficient amounts to test their antifungal activities. 1-Methoxy-3-methylindol-2-one (8) was prepared (44% yield) by catalytic hydrogenation¹⁰ of 1-methoxy-2-oxoindole-3-carboxaldehyde (21) (Scheme 7). S-Methyl 1methoxy-3-aminomethylene-2-oxindole thiocarbamate (10) was obtained upon acidic hydrolysis (dioxane-HCl) of a wasalexin mixture (1a, 1b) (65% yield, Scheme 7). Ethyl 1-methoxy-3aminomethylene-2-oxindole dithiocarbamate (20) was prepared from enamine 12 in 33% yield (based on 1-methoxyindol-2one)² using EtI as an alkylating agent (Scheme 7). S-Methyl 1methoxy-3-aminomethyl-2-oxindole thiocarbamate (11), methyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (14) and ethyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (25) were synthesized in moderate yields (11, 56%; 14, 63%; 25, 67%) using NaBH₃CN-AcOH reduction¹¹ of the corresponding compounds 10, 13 and 20 (Scheme 7). Subsequent methylation of dithiocarbamates 14 and 25 with (CH₃)₂SO₄-K₂CO₃ in acetone afforded dihydrowasalexin (7) in 51% yield and S-ethyl, S-methyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (16) in 50% yield (Scheme 7). When dithiocarbamate 25 was treated with EtI-K₂CO₃, S,S-diethyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (17) was obtained in 40% yield (Scheme 7). 1-Methoxy-3-chloromethyleneindol-2-one (22)² was used as a starting material for the synthesis of 1-methoxy-3-methylsulfanylmethylindol-2-one (9) and 1-methoxy-3-ethylsulfanylmethylindol-2-one (18). Treatment of 22 with MeSNa or EtSH-Et₃N in THF afforded sulfides 23 and 24 which were subjected to NaBH₃CN-AcOH-mediated reduction¹¹ (Scheme 7). 1-Methoxy-3-methylsulfanylmethylindol-2-one (9) and 1-methoxy-3-ethylsulfanylmethylindol-2-one (18)



Scheme 7 Synthesis of compounds 7–11, 13, 14, 16–18, 20 and 23–25.

were obtained in 26 and 23% yield, respectively, based on chloromethylene indol-2-one **22** (Scheme 7).

Antifungal bioassays

Antifungal bioassays of wasalexins 1a and 1b, methyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13) and their metabolites against L. maculans isolates BJ 125 and Mayfair 2 were carried out as summarized in Table 4 and described in the Experimental section. Wasalexins 1a and 1b and dithiocarbamate 13 were used at 2×10^{-4} M, a concentration lower than that used routinely $(5 \times 10^{-4} \text{ M})^4$ due to their low solubility in potato dextrose agar media. Wasalexin A (1a) appeared to be slightly more antifungal against isolate Mayfair 2 than to isolate BJ 125 (Table 4), whereas the antifungal activity of dihydrowasalexin (7) was higher against isolate BJ 125 than to isolate Mayfair 2. Furthermore, the antifungal activity of 1-methoxy-3-methylindol-2-one (8), the final biotransformation product of wasalexin A (1a) in isolate BJ 125 was significantly lower than wasalexin A. Interestingly, among all compounds tested for antifungal activity against isolates BJ 125 and Mayfair 2, 1-methoxy-3-methylsulfanylmethylindol-2one (9), that results from decomposition of dihydrowasalexin (7),

appeared to be the most active, completely inhibiting the growth of *L. maculans* isolate BJ 125 at 5×10^{-4} M.

Conclusion

Wasalexins A and B are phytoalexins produced by plants of the same family but rather different species that are resistant to the fungal pathogen L. maculans. To better understand the effect of these phytoalexins on the pathogen and the reactions of the pathogen to the phytoalexins, metabolic studies were carried out. Similar to other crucifer phytoalexins,³ wasalexins A (1a) and B (1b) were found to be efficiently metabolized by L. maculans isolates BJ 125 (6 h) and Mayfair 2 (<12 h). The key step in this biotransformation appeared to be the reduction of double bond in the side chain of the oxoindole ring. To compare the reactions of the iminodithioate group of wasalexins with those of dithiocarbamate groups, e.g. 2, the biotransformation of 13 in L. maculans isolates BJ 125 and Mayfair 2 was investigated as well. The metabolism of 13 in isolates BJ 125 and Mayfair 2 yielded metabolites 8 and 12, respectively. The antifungal activities of the various metabolites indicate that the metabolism of wasalexins A (1a), B (1b) and 13 in L. maculans isolates BJ 125 and Mayfair 2 are all detoxification processes, since metabolites with lower antifungal activity are obtained. Interestingly, methyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13) was reduced faster in isolate BJ 125 (major pathway) than in isolate Mayfair 2. Furthermore, perhaps significantly, this reductive transformation was substantially slower (48 h in BJ 125 and 72 h in Mayfair 2) than those of wasalexins, suggesting that the enzyme(s) is (are) somewhat specific. Although further work needs to be carried out, based on the structures of the metabolic products it is tempting to propose that similar enzymes are involved in the reduction reactions occurring in both isolates BJ 125 and Mayfair 2. On the other hand, hydrolysis of thiocarbamate 10 and dithiocarbamate 13 seems to occur only in isolate Laird 2/Mayfair 2. Since detoxification of the dithiocarbamate group of brassinin (2) via hydrolysis was previously reported in isolates of L. maculans that were weakly virulent towards canola,³ this putative hydrolase might be an enzyme characteristic of such group.

Finally it is pertinent to compare the metabolism of wasalexins A and B and dithiocarbamate 13 with that of brassinin (2) in isolates that are virulent towards canola (BJ 125 and similar group); however, no similar studies have been reported for isolate Mayfair 2. Enzymatic transformation of brassinin (2) in cultures of isolate BJ 125 yielded aldehyde 3 (Scheme 1), which is also the first product of enzymatic detoxification, apparently carried out by brassinin oxidase.¹² In addition, screening a large number of compounds related to brassinin (2) suggested that the -CH₂-NH-C=S moiety at the C-3 of indole is required for the compound to be enzymatically oxidized by brassinin oxidase.13 The work reported here is consistent with those findings and further suggests that replacement of the indole nucleus with an indol-2-one can prevent oxidation of the CH₂ attached to C-3. Considering that the transformation of dithiocarbamate 14 in cultures of isolate BJ 125 is not an oxidation and that its biotransformation is rather slow (20% remaining after 96 h), it would be important to evaluate its effect on the metabolism of brassinin (2). If reversible (or irreversible) inhibition of brassinin oxidase is observed, compound

Compound	Concentration/M	BJ 125% inhibition ^a	Laird 2/Mayfair 2% inhibition
Wasalexin A (1a)	1×10^{-4} 2×10^{-4}	$\begin{array}{c} 36\pm5\\ 43\pm3 \end{array}$	$51 \pm 6 \\ 59 \pm 5$
Dihydrowasalexin (7)	5×10^{-4} 1×10^{-4} 2×10^{-4}	NS ^b 41 \pm 8 51 \pm 5	NS ^b NI ^c 23 + 2
1-Methoxy-3-methylindol-2-one (8)	$5 \times 10^{-4} \\ 1 \times 10^{-4} \\ 2 \times 10^{-4}$	63 ± 4 NI ^c	34 ± 4 NI ^c
1-Methoxy-3-methylsulfanylmethylindol-2-one (9)	$5 \times 10^{-4} \\ 1 \times 10^{-4} \\ 2 \times 10^{-4}$	21 ± 4 50 ± 5 54 ± 4	31 ± 9 36 ± 2 42 ± 6
S-Methyl 1-methoxy-3-aminomethylene-2-oxindole thiocarbamate (10)	$ \begin{array}{c} 2 \times 10 \\ 5 \times 10^{-4} \\ 1 \times 10^{-4} \\ 2 \times 10^{-4} \end{array} $	$ \begin{array}{l} 100 \pm 0 \\ \text{NI}^{c} \\ \text{NI}^{c} \end{array} $	$ \begin{array}{l} 42 \pm 0 \\ 50 \pm 4 \\ \text{NI}^c \\ 39 \pm 4 \end{array} $
S-Methyl 1-methoxy-3-aminomethyl-2-oxindole thiocarbamate (11)	2×10^{-4} 5×10^{-4} 1×10^{-4} 2×10^{-4}	NI ^c	NS ^b
1-Methoxy-3-aminomethyleneindol-2-one (12)	2×10^{-4} 5×10^{-4} 1×10^{-4}	19 ± 3 25 ± 5 NI ^c	$\frac{N\Gamma}{29 \pm 5}$
Methyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13)	2×10^{-4} 5×10^{-4} 1×10^{-4}	NI^{c} 31 ± 4 32 ± 5 27 ± 2	NI^{c} 30 ± 8 35 ± 6
Methyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (14)	2×10^{-4} 5×10^{-4} 1×10^{-4}	37 ± 3 NS ^b 35 ± 3	$ \begin{array}{l} 44 \pm 5 \\ \text{NS}^{b} \\ 33 \pm 9 \\ 50 \pm 5 \end{array} $
1-Methoxyspirobrassinin (15)	2×10^{-4} 5×10^{-4} 1×10^{-4} 2×10^{-4}	49 ± 4 69 ± 4 28 ± 5 35 ± 7	50 ± 5 56 ± 2 38 ± 8 42 ± 10
	5×10^{-4}	55 ± 7 55 ± 4	42 ± 10 54 ± 2

 Table 4
 Results of antifungal bioassays against Leptosphaeria maculans isolates BJ 125 and Laird 2/Mayfair 2

" % inhibition = $100 - [(\text{growth on treated/growth on control}) \times 100] \pm \text{SD}$; results are the means of at least three independent experiments conducted in triplicate." NS = not soluble. "NI = no inhibition.

14 might be a good lead structure for development of a second generation of brassinin detoxification inhibitors.

Experimental

General experimental procedures

All chemicals were purchased from Sigma-Aldrich Canada Ltd., Oakville, ON. All solvents were HPLC grade and used as such, except for THF (dried over Na/benzophenone). Organic extracts were dried with Na₂SO₄ and solvents removed under reduced pressure in a rotary evaporator. Flash column chromatography (FCC) was carried out using silica gel grade 60, mesh size 230-400 Å. Preparative thin layer chromatography (PTLC) was carried out on silica gel plates, Kieselgel 60 F_{254} (20 × 20 cm × 0.25 mm), compounds were visualized under UV light. Melting points were obtained on a melting point apparatus and are uncorrected. Specific rotations $[a]_{D}$ were determined at ambient temperature on a polarimeter using a 1 ml, 10 cm path length cell; the units are 10⁻¹ deg cm² g⁻¹ and the concentrations are reported in g per 100 ml. HPLC analysis was carried out with a high performance liquid chromatograph equipped with quaternary pump, automatic injector and diode array detector (wavelength range 190-600 nm), degasser, and a reverse phase C-18 column (5 µm particle size silica, 4.6 i.d. \times 200 mm), equipped with an in-line filter. Mobile phase: 75% H₂O-25% CH₃CN to 100% CH₃CN for 35 min, linear gradient, flow rate 1 ml min⁻¹. UV spectra were recorded in CH₃CN. Fourier transform IR spectra were obtained in KBr.

NMR spectra were recorded on 500 series spectrometers (see the ESI†); for ¹H (500 MHz), δ values were referenced as follows CDCl₃ (7.23 ppm); for ¹³C (125 MHz) CDCl₃ (77.23 ppm). Mass spectra (MS) were obtained on a mass spectrometer using a solids probe.

Fungal cultures

Fungal cultures of *L. maculans* isolates BJ 125 and Mayfair 2 were obtained from the IBCN collection, Agriculture and Agri-Food Canada Research Station, Saskatoon SK. Cultures were handled as described previously.¹⁴

Fungal metabolism

Time course study. Compounds to be investigated (1a, 1b and 7–15) were dissolved in DMSO and were added to cultures of *L. maculans* isolates BJ 125 or Mayfair 2 (10⁸ spores per 100 ml) in minimal media (48-h-old, 100 ml in 250 ml Erlenmeyer flasks, final concentration 5×10^{-5} M). Cultures were incubated on a shaker at 130 rpm, 24 ± 2 °C. Samples (5 ml) were withdrawn at appropriate times, frozen or immediately extracted with EtOAc (2 × 5 ml). Organic and water phases were concentrated, residues were dissolved in CH₃CN (1 ml) and were subjected for HPLC analysis. Experiments were performed in triplicate, control flasks containing the mycelia in minimal media as well as tested compounds in minimal media were incubated under similar conditions. For biotransformations in water, 7-d-old cultures of isolates BJ 125

and Mayfair 2 (10^8 per 100 ml spores) were filtered off, the mycelia was washed with sterile water and was transferred to 100 ml of sterile water in 250 ml Erlenmeyer flasks. The compounds were added to the mycelial cultures and worked up and analyzed as described above.

Large-scale metabolism experiments. Compounds 1a, 1a + 1b, 10 and 13 were dissolved in DMSO (1.2 ml) and were added to cultures of isolates BJ 125 or Mayfair 2 (108 spores per 100 ml) in minimal media (48-h-old mycelia) or in water (7-d-old mycelia). Cultures were incubated on a shaker at 130 rpm, 24 ± 2 °C as follows: 1a, 1a + 1b, (24 h); 10 (BJ 125 48 h, Mayfair 2 96 h); 13 (72 h). The mycelia were filtered off and the filtrates were extracted with EtOAc. The combined organic extracts were dried, were concentrated and were subjected for PTLC as follows: 1a, 1a + 1b and 13 (hexane-EtOAc (2:1)); 10 (hexane-acetone (2:1)). Metabolites (listed in order of decreasing polarity) obtained from wasalexin A (1a) metabolism: in BJ 125, S-methyl 1-methoxy-3-aminomethyl-2-oxindole thiocarbamate (11, 2 mg, $[a]_{D}$ –1.5 (c 0.13, CH₂Cl₂)); 1-methoxy-3-methylindol-2-one (8, 3 mg, [a]_D 0.10 (c 0.13, CH₂Cl₂)); in Mayfair 2, S-methyl 1-methoxy-3-aminomethyl-2oxindole thiocarbamate (11, 2 mg, $[a]_{D}$ –0.50 (c 0.10, CH₂Cl₂)); 1-methoxy-3-methyl oxindole (8, 3 mg, $[a]_{D} - 4.6 (c \, 0.12, CH_2Cl_2))$. Metabolites obtained from methyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13) metabolism: in BJ 125, methyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (14, 3 mg, $[a]_{\rm D}$ 5.7 (c 0.13, CH₂Cl₂)); methyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13, 5 mg recovered, 30%); in Mayfair 2, 1-methoxy-3-aminomethyleneindol-2-one (12, 2 mg); methyl 1methoxy-3-aminomethyl-2-oxindole dithiocarbamate (14, 1 mg, $[a]_{\rm D}$ 0.9 (c 0.09, CH₂Cl₂)); methyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (13, 2 mg recovered, 12%). Metabolites obtained from S-methyl 1-methoxy-3-aminomethylene-2oxindole thiocarbamate (10) metabolism: in BJ 125, S-methyl 1methoxy-3-aminomethyl-2-oxindole thiocarbamate $(11, 2 \text{ mg}, [a]_D)$ 6.0 (c 0.12, CH₂Cl₂)); S-methyl 1-methoxy-3-aminomethylene-2oxindole thiocarbamate (10, 3 mg recovered, 19%); in Mayfair 2, 1-methoxy-3-aminomethyleneindol-2-one (12, 5 mg); S-methyl 1methoxy-3-aminomethylene-2-oxindole thiocarbamate (10, 1 mg recovered, 6%).

Antifungal bioassays

Antifungal bioassays against *L. maculans* isolates BJ 125 and Mayfair 2 were performed as follows: a DMSO solution of the compound to be tested (final concentration 5×10^{-4} , 2×10^{-4} , 1×10^{-4} M, 1% final DMSO concentration) was added to potato dextrose agar medium at *ca.* 50 °C, was mixed quickly and was poured onto 6-well plates (2.5 ml). An agar plug (8 mm diameter) cut from edges of 7-d-old solid cultures was placed upside down on the center of each plate and the plates were incubated at $24 \pm$ 2 °C under constant light for 5 d (BJ 125) or 6 d (Mayfair 2). The diameter of the mycelia (in mm) was then measured and compared with control plates containing only DMSO. Each assay was conducted in triplicate and repeated at least three times.

Time course study of decomposition of dihydrowasalexin (7) and S-ethyl,S-methyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (16) in minimal media. Compounds 7 and 16 were dissolved in DMSO (0.5 ml) and were added to minimal media

Synthesis

Dihydrowasalexin (7) and S-ethyl,S-methyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (16). Dimethyl sulfate (14 μ l, 0.15 mmol) and K₂CO₃ (21 mg, 0.15 mmol) were added to separate solutions of methyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (14, 32 mg, 0.11 mmol) or ethyl 1-methoxy-3aminomethyl-2-oxindole dithiocarbamate (25, 37 mg, 0.12 mmol) in acetone (500 μ l) and the mixtures were stirred for 20 h at rt. The reaction mixtures were diluted with brine (20 ml) and were extracted with EtOAc. The combined organic extracts were dried, were concentrated and the residues were subjected for FCC as follows: 14, hexane-acetone (5 : 1); 25, hexane-EtOAc (4 : 1) to afford dihydrowasalexin (7, 17 mg, 51%) and S-ethyl,S-methyl 1-methoxy-3-aminomethyl-2-oxindole iminodithioate (16, 19 mg, 50%, mixture of E : Z isomers, 1 : 1).

Dihydrowasalexin (7). Slightly yellow solid, mp 79–81 °C; HPLC: $t_{\rm R} = 23.7$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 232 (log ε, 3.9); $v_{\rm max}$ (KBr)/cm⁻¹: 2924, 1730, 1613, 1578, 1466, 1318, 1232, 1034, 749. $\delta_{\rm H}$ (500 MHz; CDCl₃): 2.18 (s, 3H), 2.53 (s, 3H), 3.75 (m, 1H), 3.86 (dd, J = 6.5, 6.5 Hz, 1H), 4.02 (m, 1H), 4.04 (s, 3H), 6.98 (d, J = 7.5 Hz, 1H), 7.07 (dd, J = 7.5, 7.5 Hz, 1H), 7.31 (m, 2H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 14.9 (2 × q), 45.6 (d), 52.8 (t), 63.6 (q), 107.0 (d), 123.0 (d), 125.0 (d), 125.1 (s), 128.3 (d), 141.1 (s), 154.7 (s), 171.3 (s). HRMS (EI): calc. for C₁₃H₁₆N₂O₂S₂ (M⁺) m/z 296.0653, found 296.0655. MS (EI) m/z (% relative abundance): 296 [M⁺] (22), 249 (68), 176 (68), 134 (100), 117 (48).

S-*Ethyl*,S-*methyl* 1-*methoxy-3-aminomethyl-2-oxindole iminodithioate* (16). Slightly yellow oil, mixture of *E* and *Z* isomers; HPLC: $t_{\rm R} = 26.4$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 232 (log ε , 3.7); $\nu_{\rm max}$ (KBr)/cm⁻¹: 2926, 1727, 1618, 1580, 1464, 1322, 1228, 1034, 749. $\delta_{\rm H}$ (500 MHz; CDCl₃): 1.00 (t, *J* = 7.5 Hz, 3H), 1.34 (t, *J* = 7.5 Hz, 3H), 2.19 (s, 3H), 2.51 (s, 3H), 2.70 (m, 1H), 2.82 (m, 1H), 3.07 (q, *J* = 7.5 Hz, 2H), 3.74 (m, 1H), 3.90 (m, 1H), 4.04 (m, 1H), 4.04 (s, 3H), 6.98 (d, *J* = 8 Hz, 1H), 7.06 (dd, *J* = 8, 8 Hz, 1H), 7.30 (m, 2H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 14.1 (q), 14.7 (q), 14.9 (q), 15.4 (q), 25.9 (t), 26.7 (t), 45.6 (2 × d), 52.7 (t), 53.0 (t), 63.6 (2 × q), 107.0 (2 × d), 123.0 (2 × d), 124.8 (d), 125.0 (d), 125.1 (s), 125.2 (s), 128.3 (d), 141.1 (s), 141.2 (s), 160.2 (br s), 171.3 (s), 171.4 (s). HRMS (EI): calc. for C₁₄H₁₈N₂O₂S₂ (M⁺) *m/z* 310.0810, found 310.0812. MS (EI) *m/z* (% relative abundance): 310 [M⁺] (24), 263 (62), 176 (100), 148 (79), 117 (91).

1-Methoxy-3-methylindol-2-one (8). 10% Pd/C (45 mg) was added to a solution of 1-methoxy-2-oxoindole-3-carboxaldehyde² (**21**, 96 mg, 0.5 mmol) in EtOH (7 ml). The mixture was shaken at 3 atm of H₂ for 5 min at rt. The catalyst was filtered off, the filter was washed with CH₂Cl₂ and the filtrate was concentrated. The residue was subjected to FCC, hexane–acetone (5 : 1), to afford 1-methoxy-3-methylindol-2-one (**8**, 39 mg, 44%) as colorless solid, mp 62–64 °C; HPLC: $t_{\rm R} = 10.3$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 238 (log ε , 3.6); $v_{\rm max}$ (KBr)/cm⁻¹: 2979, 2938, 1727, 1618, 1466, 1323, 1239, 1101, 1051, 963, 748. $\delta_{\rm H}$ (500 MHz; CDCl₃): 1.52 (d, J = 7.5 Hz, 3H), 3.45 (q, J = 7.5 Hz, 1H), 4.05 (s, 3H), 7.00 (d, J = 7.5 Hz,

1H), 7.10 (dd, J = 7.5, 7.5 Hz, 1H), 7.26 (d, J = 7.5 Hz, 1H), 7.32 (dd, J = 7.5, 7.5 Hz, 1H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 15.2 (q), 39.2 (d), 63.7 (q), 107.4 (d), 123.2 (d), 124.1 (d), 127.1 (s), 128.3 (d), 140.4 (s), 173.3 (s). HRMS (EI): calc. for C₁₀H₁₁NO₂ (M⁺) m/z 177.0790, found 177.0789. MS (EI) m/z (% relative abundance): 177 [M⁺] (47), 149 (100), 118 (52), 91 (19), 77 (11).

S-Methyl 1-methoxy-3-aminomethylene-2-oxindole thiocarbamate (10). HCl (0.5 M, 3 ml) was added to a solution of wasalexins⁶ (1a, 1b, 144 mg, 0.49 mmol) in dioxane (5 ml). The mixture was refluxed with stirring for 3 h, was allowed to cool (rt) and was diluted with brine (50 ml). The mixture was extracted (EtOAc), the combined extract was washed with 10% K_2CO_3 solution (20 ml), was dried and was concentrated. The residue was subjected to FCC, hexane-acetone (2:1), to yield Smethyl 1-methoxy-3-aminomethylene-2-oxindole thiocarbamate (10, 84 mg, 65%) as a slightly yellow solid. Mp 123–125 °C; HPLC: $t_{\rm R} = 20.6 \text{ min}; \lambda_{\rm max} (\rm CH_3 CN)/\rm nm \ 261 \ (\log \varepsilon, 4.4), \ 306 \ (4.4); \ \nu_{\rm max}$ (KBr)/cm⁻¹: 3266, 3050, 2942, 1687, 1630, 1458, 1236, 1116, 1052, 945, 735. $\delta_{\rm H}$ (500 MHz; CDCl₃): 2.51 (s, 3H), 4.08 (s, 3H), 7.04 (d, J = 8 Hz, 1H), 7.10 (dd, J = 8, 8 Hz, 1H), 7.28 (dd, J = 8, 8 Hz, 1H), 7.42 (d, J = 8 Hz, 1H), 8.07 (d, J = 11 Hz, 1H), 10.88 (br d, D₂O exch., J = 11 Hz, 1H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 12.9 (q), 64.3 (q), 104.1 (s), 107.6 (d), 118.7 (s), 118.7 (d), 122.9 (d), 127.6 (d), 131.2 (d), 136.7 (s), 163.9 (s), 168.4 (s). HRMS (EI): calc. for $C_{12}H_{12}N_2O_3S$ (M⁺) m/z 264.0569, found 264.0562. MS (EI) m/z(% relative abundance): 264 [M⁺] (100), 233 (16), 217 (45), 185 (21), 157 (63), 130 (16), 103 (11).

Preparation of compounds 9, 11, 14, 18 and 25. NaBH₃CN (415 mg, 6.6 mmol) was added to separate solutions of compounds **10, 13, 20, 23** and **24** (0.33 mmol) in AcOH (4 ml). The mixtures were stirred at 55 °C for 18 h, were diluted with brine (30 ml) and were extracted (EtOAc). The combined organic extracts were washed with $10\% K_2CO_3$ solution (30 ml), were dried, were concentrated and the residues were subjected to FCC as follows: **10** and **13**, hexane–acetone (2 : 1); **20**, hexane–acetone (5 : 1); **23** and **24**, hexane–EtOAc (4 : 1) to yield the following products.

1-Methoxy-3-methylsulfanylmethylindol-2-one (**9**, 29 mg, 26%, based on **22**), slightly yellow oil. HPLC: $t_{\rm R} = 14.1$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 238 (log ε , 3.7); $v_{\rm max}$ (KBr)/cm⁻¹: 2924, 1726, 1618, 1465, 1325, 1225, 1076, 750. $\delta_{\rm H}$ (500 MHz; CDCl₃): 2.97 (dd, J = 7.5, 7.5 Hz, 1H), 3.28 (dd, J = 4, 4 Hz, 1H), 3.67 (dd, J = 7.5, 4 Hz, 1H), 4.05 (s, 3H), 7.02 (d, J = 8 Hz, 1H), 7.11 (dd, J = 8, 8 Hz, 1H), 7.36 (dd, J = 8, 8 Hz, 1H), 7.49 (d, J = 8 Hz, 1H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 16.9 (q), 35.2 (d), 44.2 (t), 63.7 (q), 107.5 (d), 123.3 (d), 124.6 (s), 125.2 (d), 128.8 (d), 140.9 (s), 170.9 (s). HRMS (EI): calc. for C₁₁H₁₃NO₂S (M⁺) *m*/*z* 223.0667, found 223.0661. MS (EI) *m*/*z* (% relative abundance): 223 [M⁺] (79), 192 (38), 164 (47), 148 (67), 133 (16), 117 (100), 90 (33).

S-Methyl 1-methoxy-3-aminomethyl-2-oxindole thiocarbamate (11, 50 mg, 56%), colorless solid. Mp 115–118 °C; HPLC: $t_{\rm R} =$ 9.0 min; $\lambda_{\rm max}$ (CH₃CN)/nm 238 (log ε , 3.7); $\nu_{\rm max}$ (KBr)/cm⁻¹: 3300, 2939, 1714, 1615, 1525, 1466, 1321, 1226, 1082, 747. $\delta_{\rm H}$ (500 MHz; CDCl₃): 2.37 (s, 3H), 3.46 (m, 1H), 3.62 (m, 1H), 4.05 (s, 3H), 4.19 (m, 1H), 6.27 (br s, D₂O exch., 1H), 7.02 (d, J = 8 Hz, 1H), 7.13 (dd, J = 8, 8 Hz, 1H), 7.33 (d, J = 8 Hz, 1H), 7.36 (dd, J = 8, 8 Hz, 1H), 7.33 (d, J = 8 Hz, 1H), 7.36 (dd, J = 8, 8 Hz, 1H), 7.33 (d, J = 8 Hz, 1H), 7.36 (dd, J = 8, 8 Hz, 1H), 7.33 (d, J = 8 Hz, 1H), 7.36 (dd, J = 8, 8 Hz, 1H), 7.33 (d, J = 8 Hz, 1H), 7.36 (dd, J = 8, 8 Hz, 1H), 7.36 (d), 122.3 (s), 123.7 (d), 124.8 (d), 129.2 (d), 140.8 (s), 168.7 (s), 171.6 (s). HRMS (EI): calc. for C₁₂H₁₄N₂O₃S (M⁺) *m*/*z*

266.0725, found 296.0727. MS (EI) *m*/*z* (% relative abundance): 266 [M⁺] (12), 175 (100), 144 (31), 117 (19).

Methyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (14, 59 mg, 63%), slightly yellow solid. Mp 112–114 °C; HPLC: $t_{\rm R} = 18.5$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 238 (log ε , 4.2); $v_{\rm max}$ (KBr)/cm⁻¹: 3246, 2917, 1699, 1617, 1509, 1465, 1319, 1103, 951, 748. $\delta_{\rm H}$ (500 MHz; CDCl₃): 2.67 (s, 3H), 3.57 (m, 1H), 3.85 (m, 1H), 4.06 (s, 3H), 4.88 (m, 1H), 7.04 (d, J = 7.5 Hz, 1H), 7.15 (dd, J = 7.5, 7.5 Hz, 1H), 7.38 (m, 2H), 8.22 (br s, D₂O exch., 1H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 18.3 (q), 42.4 (d), 46.5 (t), 64.0 (q), 107.9 (d), 121.9 (s), 123.9 (d), 124.9 (d), 129.3 (d), 140.6 (s), 172.0 (s), 200.2 (s). HRMS (EI): calc. for C₁₂H₁₄N₂O₂S₂ (M⁺) *m/z* 282.0497, found 282.0505. MS (EI) *m/z* (% relative abundance): 282 [M⁺] (10), 251 (61), 234 (83), 175 (100), 144 (75), 117 (57), 63 (11).

1-Methoxy-3-ethylsulfanylmethylindol-2-one (**18**, 28 mg, 23%, based on **22**), slightly yellow oil. HPLC: $t_{\rm R} = 17.4$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 238 (log ε , 3.8); $v_{\rm max}$ (KBr)/cm⁻¹: 2927, 1726, 1617, 1462, 1324, 1217, 1068, 750. $\delta_{\rm H}$ (500 MHz; CDCl₃): 1.25 (t, J = 7.5 Hz, 3H), 2.58 (q, J = 7.5 Hz, 2H), 2.97 (dd, J = 7.5, 7.5 Hz, 1H), 3.33 (dd, J = 3.5, 3.5 Hz, 1H), 3.65 (m, 1H), 4.05 (s, 3H), 7.02 (d, J = 7.5 Hz, 1H), 7.11 (dd, J = 7.5, 7.5 Hz, 1H), 7.35 (dd, J = 7.5, 7.5 Hz, 1H), 7.49 (d, J = 7.5 Hz, 1H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 15.0 (q), 27.3 (t), 32.6 (d), 44.4 (t), 63.7 (q), 107.5 (d), 123.3 (d), 124.6 (s), 125.2 (d), 128.8 (d), 140.9 (s), 170.9 (s). HRMS (EI): calc. for C₁₂H₁₅NO₂S (M⁺) *m*/*z* 237.0824, found 237.0823. MS (EI)*m*/*z* (% relative abundance): 237 [M⁺] (45), 206 (16), 175 (11), 148 (24), 117 (33), 90 (12).

Ethyl 1-methoxy-3-aminomethyl-2-oxindole dithiocarbamate (25, 66 mg, 67%), slightly yellow solid. Mp 134–136 °C; HPLC: $t_{\rm R} = 21.1$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 238 (log ε , 4.1); $v_{\rm max}$ (KBr)/cm⁻¹: 3263, 2927, 1712, 1619, 1507, 1462, 1323, 1091, 948, 749. $\delta_{\rm H}$ (500 MHz; CDCl₃): 1.38 (t, J = 7.5 Hz, 3H), 3.28 (q, J = 7.5 Hz, 2H), 3.57 (m, 1H), 3.85 (m, 1H), 4.06 (s, 3H), 4.87 (m, 1H), 7.04 (d, J = 8 Hz, 1H), 7.15 (dd, J = 8, 8 Hz, 1H), 7.38 (m, 2H), 8.15 (br s, D₂O exch., 1H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 14.2 (q), 29.8 (t), 42.4 (d), 46.3 (t), 64.0 (q), 107.9 (d), 122.0 (s), 123.9 (d), 124.9 (d), 129.3 (d), 140.7 (s), 172.0 (s), 199.5 (s). HRMS (EI): calc. for C₁₃H₁₆N₂O₂S₂ (M⁺) *m*/*z* 296.0653, found 296.0659. MS (EI) *m*/*z* (% relative abundance): 296 [M⁺] (10), 265 (100), 234 (43), 203 (13), 175 (92), 145 (21), 117 (65), 77 (19).

1-Methoxy-3-aminomethyleneindol-2-one (12). For the synthesis and spectral data (IR, NMR, MS) see reference.² λ_{max} (CH₃CN)/nm 258 (log ε , 4.3), 294 (4.2).

1-Methoxyspirobrassinin (15). For synthesis see reference;⁷ for spectral data see reference.¹⁵

S,*S*-Diethyl 1-methoxy-3-aminomethylene-2-oxindole iminodithioate (17). K₂CO₃ (26 mg, 0.19 mmol) and EtI (30 µl, 0.37 mmol) were added to a stirred solution of ethyl 1-methoxy-3-aminomethylene-2-oxindole dithiocarbamate (25, 46 mg, 0.16 mmol) in acetone. The mixture was stirred for 18 h at rt, was diluted with brine (25 ml) and was extracted with EtOAc. The combined organic extract was dried, was concentrated and the residue was subjected to FCC, hexane–EtOAc (4 : 1), to afford *S*,*S*diethyl 1-methoxy-3-aminomethylene-2-oxindole iminodithioate (17, 20 mg, 40%) as a slightly yellow oil. HPLC: $t_{\rm R} = 29.4$ min; $\lambda_{\rm max}$ (CH₃CN)/nm 232 (log ε , 3.7); $v_{\rm max}$ (KBr)/cm⁻¹: 2928, 1727, 1618, 1579, 1463, 1322, 1228, 1084, 749. $\delta_{\rm H}$ (500 MHz; CDCl₃): 1.03 (t, J = 7 Hz, 3H), 1.35 (t, J = 7 Hz, 3H), 2.75 (m, 1H), 2.88 (m, 1H), 3.06 (q, J = 7 Hz, 2H), 3.77 (m, 1H), 3.94 (dd, J = 5.5, 5.5 Hz, 1H), 4.04 (s, 3H), 4.07 (m, 1H), 6.98 (d, J = 8 Hz, 1H), 7.07 (dd, J = 8, 8 Hz, 1H), 7.30 (m, 2H). $\delta_{\rm C}$ (125 MHz; CDCl₃): 14.1 (q), 15.3 (q), 26.1 (t), 26.8 (t), 45.5 (d), 52.6 (t), 63.6 (q), 107.1 (d), 123.1 (d), 124.9 (d), 125.0 (s), 128.3 (d), 141.2 (s), 152.4 (s), 171.4 (s). HRMS (EI): calc. for C₁₅H₂₀N₂O₂S₂ (M⁺) *m/z* 324.0966, found 324.0973. MS (EI) *m/z* (% relative abundance): 324 [M⁺] (15), 263 (93), 176 (100), 148 (64), 117 (69).

Ethyl 1-methoxy-3-aminomethylen-2-oxindole dithiocarbamate (20). CS_2 (72 µl, 1.2 mmol) and NaH (60% suspension in mineral oil, 33 mg, 1 mmol) were added to a stirred, cooled (0 °C) solution of 1-methoxy-3-aminomethylene oxindole (12, prepared from 1 mmol of 1-methoxyoxindole⁶) in dry THF (2 ml). After 5 min at 0 °C, EtI (88 µl, 1.1 mmol) was added and the stirring continued for further 1 h at 0 °C. The mixture was diluted with 1 M HCl (30 ml) and was extracted with EtOAc. The combined organic extract was dried, was concentrated and the residue was subjected for FCC, hexane-acetone (5 : 1). Ethyl 1-methoxy-3aminomethylene-2-oxindole dithiocarbamate (20, 97 mg, 33%, based on 1-methoxyoxindole) was obtained as a bright yellow solid, mp. 123–125 °C; HPLC: $t_{\rm R} = 34.3 \text{ min}; \lambda_{\rm max} (\rm CH_3 CN)/nm$ 269 (log ε , 4.2), 348 (4.3); v_{max} (KBr)/cm⁻¹: 3178, 2932, 1684, 1641, 1496, 1454, 1217, 1101, 1056, 996, 915, 778, 742. $\delta_{\rm H}$ (500 MHz; $CDCl_3$): 1.44 (t, J = 7.5 Hz, 3H), 3.39 (q, J = 7.5 Hz, 2H), 4.10 (s, 3H), 7.05 (d, *J* = 8 Hz, 1H), 7.12 (dd, *J* = 8, 8 Hz, 1H), 7.30 (dd, J = 8, 8 Hz, 1H), 7.50 (d, J = 8 Hz, 1H), 8.68 (d, J = 10 Hz, 1H), 12.10 (br d, D₂O exch., J = 10 Hz, 1H). δ_{C} (125 MHz; CDCl₃): 13.6 (q), 30.5 (t), 64.4 (q), 106.1 (s), 107.8 (d), 118.6 (s), 119.3 (d), 123.1 (d), 128.0 (d), 133.0 (s), 137.0 (d), 163.9 (s), 199.4 (s). HRMS (EI): calc. for $C_{13}H_{14}N_2O_2S_2$ (M⁺) m/z 294.0497, found 294.0498. MS (EI) m/z (% relative abundance): 294 [M⁺] (65), 263 (100), 232 (27), 201 (19), 173 (79), 144 (23).

1-Methoxy-3-methylsulfanylmethyleneindol-2-one (23). MeSNa (88 mg, 1.25 mmol) was added to a solution of 1-methoxy-3-chloromethyleneindol-2-one² (22, 105 mg, 0.5 mmol) in THF (3 ml). The mixture was stirred for 2 h at rt, was concentrated, the residue was suspended in CH_2Cl_2 (10 ml) and SiO₂ was added. The residue, obtained after concentration was sub-

jected to FCC, hexane-acetone (2:1), to afford 1-methoxy-3-methylsulfanylmethyleneindol-2-one (23, 74 mg, 67%) as yellow oil, which was immediately used in the next step.

1-Methoxy-3-ethylsulfanylmethyleneindol-2-one (24). EtSH (92 μ l, 1.25 mmol) and Et₃N (170 μ l, 1.25 mmol) were added to a solution of 1-methoxy-3-chloromethyleneindol-2-one² (22, 105 mg, 0.5 mmol) in THF (3 ml). The mixture was stirred for 15 min at rt, was diluted with brine (20 ml) and was extracted with EtOAc. The combined extract was dried, was concentrated and the residue was subjected to FCC, hexane–acetone (5 : 1), to afford 1-methoxy-3-methylsulfanylmethyleneindol-2-one (24, 87 mg, 74%) as yellow oil, which was immediately used in the next step.

References

- 1 M. S. C. Pedras, J. L. Sorensen, F. I. Okanga and I. L. Zaharia, *Bioorg. Med. Chem. Lett.*, 1999, 9, 3015.
- 2 M. S. C. Pedras, P. B. Chumala and M. Suchý, *Phytochemistry*, 2003, **64**, 949.
- 3 For a review of phytoalexin detoxification in plant pathogenic fungi: M. S. C. Pedras and P. W. K. Ahiahonu, *Phytochemistry*, 2005, 66, 391.
- 4 M. S. C. Pedras and M. Suchý, Org. Biomol. Chem., 2005, 3, 2002.
- 5 J. L. Taylor, M. S. C. Pedras and V. M. Morales, *Trends Microbiol.*, 1995, **3**, 202.
- 6 M. S. C. Pedras and M. Jha, J. Org. Chem., 2005, 70, 1828.
- 7 M. S. C. Pedras, M. Suchý and P. W. K. Ahiahonu, Org. Biomol. Chem., 2006, 4, 691.
- 8 1-Methoxy-3-methyleneindol-2-one has not been reported but 3methyleneindol-2-one, the corresponding NH analogue, has been reported. For recent work on syntheses of 3-methyleneindol-2-ones see: (a) S. Rossiter, *Tetrahedron Lett.*, 2002, **43**, 4671; (b) S. E. V. Bell, R. F. C. Brown, F. W. Eastwood and J. M. Horvath, *Aust. J. Chem.*, 2000, **53**, 183.
- 9 L. K. Folkes, S. Rossiter and P. Wardman, *Chem. Res. Toxicol.*, 2002, 15, 877.
- 10 T. Sakan, K. Kusuda and T. Miwa, Bull. Chem. Soc. Jpn., 1964, 37, 1681.
- 11 For a review of reactions using sodium borohydrides in carboxylic acid media see: G. W. Gribble, *Chem. Soc. Rev.*, 1998, 27, 395.
- 12 M. S. C. Pedras, M. Jha and O. G. Okeola, *Phytochemistry*, 2005, 66, 2609.
- 13 M. S. C. Pedras and M. Jha, Bioorg. Med. Chem., 2006, 14, 4958.
- 14 M. S. C. Pedras and A. Q. Khan, J. Agric. Food Chem., 1996, 44, 3403.
- 15 M. S. C. Pedras and P. W. K. Ahiahonu, J. Chem. Ecol., 2004, 30, 2163.